

Acidi	Basi coniugate	Ka	pKa	range di pH
acido ftalico (C <sub>6</sub> H <sub>4</sub> (COOH) <sub>2</sub> )	ftalato (C <sub>6</sub> H <sub>4</sub> COOH(COO <sup>-</sup> ))	1.10E-03	2.96	1.96 - 3.96
acido formico (HCCOH)	formiato (HCOO <sup>-</sup> )	1.80E-04	3.74	2.74 - 4.74
acido nitroso (HNO <sub>2</sub> )	Nitrito (NO <sub>2</sub> <sup>-</sup> )	7.10E-04	3.15	2.15 - 4.15
acido acetico (CH <sub>3</sub> COOH)	acetato (CH <sub>3</sub> COO <sup>-</sup> )	1.80E-05	4.74	3.74 - 5.74
ione diidrogenofosfato (H <sub>2</sub> PO <sub>4</sub> <sup>-</sup> )	ione idrogenofosfato (HPO <sub>4</sub> <sup>2-</sup> )	6.20E-08	7.21	6.21 - 8.21
acido ipocloroso (HClO)	ipoclorito (ClO <sup>-</sup> )	3.00E-08	7.52	6.52 - 8.52
ione ammonio (NH <sub>4</sub> <sup>+</sup> )	ammoniaca (NH <sub>3</sub> )	5.50E-10	9.26	8.26 - 10.26
acido cianidrico (HCN)	cianuro (CN <sup>-</sup> )	6.20E-10	9.21	8.21 - 10.21